

THE AMERICAN COLLEGE, MADURAI

(An Autonomous Institution Affiliated to Madurai Kamaraj University) Re-accredited (2nd Cycle) by NAAC with Grade "A", CGPA – 3.46 on a 4-point scale

Backlog Arrear Examination, March 2021

| CHE/CHS 2502/PHS 2502PRINCIPLES OF CHEMISTRYTime: 3 HoursMax Marks: | | | |
|---|---|---|--|
| | Answer any FIVE Questions (5 X | X 15 = 75) | |
| 1. | a) Discuss about the temperature dependance of rate reaction by using A | perature dependance of rate reaction by using Arrhenius | |
| | equation. | | |
| | b) Derive the expression for the rate constant of first order reaction. | [7+8] | |
| 2. | a) Discuss Hess's law and its applications. | | |
| | b) How enthalpy of combustion is measured by using Bomb Calorimeter | ? [7+8] | |
| 3. | a) Discuss about the structure of NaCl. | | |
| | b) Explain Schottky and Frenkel defect. | [7+8] | |
| 4. | a) What are meso compounds? Are the meso compounds optically active | | |
| | or inactive? Explain your answer through a suitable example. | | |
| | b) Comment on the physical and chemical properties of enantiomers and | 1 | |
| | diastereomers. | [8+7] | |
| 5. | a) Discuss about Thin layer Chromatography. | | |
| | b) Explain the principle behind Paper Chromatography. | [7+8] | |
| 6. | a) Write note on Band theory. | | |
| | b) Define half life time of a reaction. How the temperature effect the rate | of | |
| | the reaction? | [7+8] | |
| 7. | a) Explain Kirchoff's equation. | | |
| | b) Write note on specific rotation. | [8+7] | |
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