

## THE AMERICAN COLLEGE, MADURAI

(An Autonomous Institution Affiliated to Madurai Kamaraj University) Re-accredited (2<sup>nd</sup> Cycle) by NAAC with Grade "A", CGPA – 3.46 on a 4-point scale

## Backlog Arrear Examination, March 2021

CHS 1425/ CHS 1371 CHEMISTRY FOR BIOCHEMIST-I Max: 75 Marks

Time: 3 Hrs

**Answer any FIVE questions** 

5 x 15=75

- 1. With neat sketch explain the Born-Haber cycle for the formation of NaCl molecule.
- 2. Distinguish  $S_N^1$  and  $S_N^2$  reactions with suitable examples.
- 3. State Le Chatelier's principle. Using that explain the effect of temperature on the following equilibrium  $N_2(g) + 3H_2(g) \leftrightarrow 2NH_3$   $\Delta H$ =-92.38 KJ
- 4. a) Write short note on the following terms
  - (i) Osmosis (ii) Osmotic pressure
  - b) Explain Reverse osmosis process with suitable diagram.
- 5. a) Differentiate nuclear fission and fusion reactions.
  - b) Write the applications of radioactivity.
- 6. Enumerate the applications of thermodynamics in biological systems.
- 7. Draw and explain the MO diagram for  $O_2$  molecule.