

**THE AMERICAN COLLEGE, MADURAI** (An Autonomous Institution Affiliated to Madurai Kamaraj University) Re-accredited (2<sup>nd</sup> Cycle) by NAAC with Grade "A", CGPA – 3.46 on a 4-point scale

## Backlog Arrear Examination, March 2021

| CHE 2503/1415                                 | Fundamentals of Chemistry-I                        | Time: 3 hours<br>Max: 75 Marks<br>(5 x 15 = 75) |        |
|---|--|---|--------|
|   | (Answer any Five Questions)                        |   |        |
| 1. a) What are quantum                        | n numbers? Write the four quantum numbers f        | or the electrons in so                          | odium  |
| (z=11).                                       |  |   |        |
| b) Define hydrogen l                          | bonding. How is it classified? Discuss the sign    | ificance of hydrogen                            | 1      |
| bonding.                                      |  |   | (6+9)  |
| 2. a) Define the follow                       | ving terms with an example i) isotope ii) isoba    | ar iii) electron affinity                       | у.     |
| b) Give the postulate molecules.              | es of VSEPR theory. Explain the geometry of        | water and ammonia                               | (6+9)  |
| 3. a) State and explain                       | the Le Chatelier principle with examples.          |   |        |
| b) Explain the term                           | osmotic pressure. How will you measure the         | osmotic pressure?                               | (7+8)  |
| 4. a) Mention the criter                      | ria for spontaneous and non-spontaneous react      | ion.  |        |
| b) Explain the entrop                         | py change for the phase transformation of the      | substance.                                      | (7+8)  |
| 5. What are the types of                      | of organic reactions? Write the following type     | s in details                                    |        |
| i) Substitution reacti                        | on ii) Addition reaction iii) Elimination reaction | on  |        |
| iv) Oxidation reactions v) Reduction reaction |  |   | (15)   |
| 6. a) Write a brief note                      | on carbocation, carbanion and free radicals.       |   |        |
| b) Differentiate the t                        | thermal and photochemical reactions.               |   | (9+6)  |
| 7. Write a short note of                      | on i) fluorescence ii) phosphorescence ii          | i) chemiuminescence                             | e (15) |